Material Safety Data Sheet

Product Name: HYDRATED LIME

INFOTRAC: 800-535-5053  [In case of an emergency call this number 24 HOURS a day 7 DAYS a week.]

1. IDENTIFICATION OF THE SUBSTANCE AND COMPANY

1.1. Identification of the substance:

Chemical name: Calcium hydroxide
Product name(s): Hydrated Lime, Industrial Hydrate
Formula: Ca(OH)₂
CAS #: 1305-62-0
Molecular Weight: 74.08
Material Uses: Water treatment, steel flux, caustic agent, pH adjustment, acid gas absorption, construction

1.2. Company:

Main Office:
11 Stanwix Street, 21st Floor
Pittsburgh, PA 15222
Telephone: 412-995-5500
Fax: 412-995-5594

Canadian Office:
P.O. Box 190
Ingersoll, Ontario N5C 3K5
Telephone: 519-423-6283
Fax: 519-423-6545

2. COMPOSITION / INFORMATION ON INGREDIENTS

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>% by Weight</th>
<th>CAS #</th>
<th>Exposure Limits</th>
</tr>
</thead>
<tbody>
<tr>
<td>Calcium hydroxide</td>
<td>&gt;85</td>
<td>1305-62-0</td>
<td>OSHA PEL: 15 mg/m³ (total), 5 mg/m³ (resp)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td>ACGIH TLV: 5 mg/m³</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td>O. Reg. 833 TWAEV: 5 mg/m³</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td>LD₅₀ oral (rat) 7340 mg/kg</td>
</tr>
<tr>
<td>Silica - crystalline quartz</td>
<td>&lt;1</td>
<td>14808-60-7</td>
<td>OSHA PEL*: 10 mg/m³ (total dust); 3.3 mg/m³ (respirable)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td>ACGIH TLV: 0.025 mg/m³ (respirable)</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td>O. Reg. 845: 0.1 mg/m³</td>
</tr>
</tbody>
</table>

*PEL (total dust) = (30 mg/m³) / (% silica + 2); PEL (respirable) = (10 mg/m³) / (% silica + 2)
### 3. HAZARDS IDENTIFICATION AND CLASSIFICATION

**Overview:**
Hydrated lime is an odorless white or grayish-white granular powder. Contact can cause irritation to eyes, skin, respiratory system, and gastrointestinal tract. Contact may aggravate disorders of eyes, skin, gastrointestinal tract, and respiratory system.

**Eyes:**
Can cause severe irritation or burning of eyes, including permanent damage.

**Skin:**
Can cause severe irritation or burning of skin, especially in the presence of moisture.

**Ingestion:**
Can cause severe irritation or burning of gastrointestinal tract if swallowed.

**Inhalation:**
Can cause severe irritation of the respiratory system. Long-term exposure may cause permanent damage. Hydrated lime is not listed by MSHA, OSHA, or IARC as a carcinogen, but this product may contain crystalline quartz silica, which has been classified by IARC as (Group I) carcinogenic to humans when inhaled. Inhalation of silica can also cause a chronic lung disorder, silicosis.

**Irritant:**
Eyes, mucous membranes, moist skin, respiratory tract.

**Flammability:**
This product is not flammable or combustible

**Explosive:**
This product is not explosive in dust form

**Reactivity:**
May react violently with strong acids producing heat and possible steam explosion in confined space

**Symbols:**
WHMIS Symbol: “E” Corrosive Material; “D2A” Materials causing other toxic effects

### 4. HEALTH EFFECTS AND TREATMENTS

**Health Effects:**

<table>
<thead>
<tr>
<th>Health Effects</th>
<th>Acute</th>
<th>Chronic</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Inhalation</strong></td>
<td>irritation, sore throat, cough, sneezing.</td>
<td>persistent coughing and breathing problems. Long-term exposure to silica can cause a chronic lung disorder, silicosis.</td>
</tr>
<tr>
<td><strong>Eyes</strong></td>
<td>severe irritation, intense tearing, burns.</td>
<td>possible blindness when exposure is prolonged.</td>
</tr>
<tr>
<td><strong>Skin</strong></td>
<td>removes natural skin oils, blotches, itching and superficial burns in case of sweating.</td>
<td>no known effects.</td>
</tr>
<tr>
<td><strong>Ingestion</strong></td>
<td>sore throat, stomach aches, cramps, diarrhea, vomiting.</td>
<td>no known effects.</td>
</tr>
</tbody>
</table>

**Treatments:**

<table>
<thead>
<tr>
<th>Treatments</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Inhalation</strong></td>
<td>Move victim to fresh air. Seek medical attention if necessary. If breathing has stopped, give artificial respiration.</td>
</tr>
<tr>
<td><strong>Eyes</strong></td>
<td>Immediately flush eyes with large amounts of water for at least 15 minutes. Pull back the eyelid to make sure all the lime dust has been washed out. Seek medical attention immediately. Do not rub eyes.</td>
</tr>
<tr>
<td><strong>Skin</strong></td>
<td>Flush exposed area with large amounts of water. Seek medical attention immediately.</td>
</tr>
<tr>
<td><strong>Ingestion</strong></td>
<td>Give large quantities of water or fruit juice. Do not induce vomiting. Seek medical attention immediately. Never give anything by mouth if victim is rapidly losing consciousness or is unconscious or convulsing.</td>
</tr>
</tbody>
</table>
### 5. FIRE FIGHTING MEASURES

- **Flash point:** Non-flammable
- **Autoignition temperature:** Non-flammable
- **Inflammability limits:** None
- **Explosion risk:** None by itself, but heat produced by reaction with strong acids can generate steam and pressure
- **Hazardous combustion products:** Decomposes to produce calcium oxide (CaO), which can react with water to produce steam and pressure
- **Extinguishing media:** Use dry chemical fire extinguisher. Do not use water or halogenated compounds, except that large amounts of water may be used to deluge small quantities of hydrated lime. Use appropriate extinguishing media for surrounding fire conditions.
- **Fire fighting instructions:** Keep personnel away from and upwind of fire. Wear full fire-fighting turn-out gear (full Bunker gear), and respiratory protection (self-contained breathing apparatus).

### 6. ACCIDENT PREVENTION MEASURES

- **Individual and collective precautions:** Avoid creating conditions which release dust – use mechanical ventilation to remove dust from work spaces
- **Avoid inhalation of dust:** Wear respiratory protection - minimum NIOSH N-95 Dust Mask
- **Cleaning methods for spills:** Use personal protective equipment (eyes, skin and inhalation, see Section 8). Use dry methods (vacuuming, sweeping) to collect spilled materials. Avoid generating dust. For large spills, evacuate area downwind of clean-up area operations to minimize dust exposure. For small spills, store spilled materials in dry, sealed plastic or metal containers. Dust residue on surfaces may be washed with water.
- **Precautions for the protection of the environment:** May not be released into surface waters without controls (increases pH)
- **Waste Disposal:** Dispose according to federal, provincial/state and local environmental regulations

### 7. HANDLING AND STORAGE

- **Handling:** In open air or in ventilated places, avoid skin and eye contact, avoid creating airborne dust
- **Storage:** Store in dry places sheltered from humidity. Keep away from acids and incompatible substances. Keep out of reach of children
8. EXPOSURE CONTROL / PERSONAL PROTECTION

Exposure Limits: Calcium hydroxide: 15 mg/m³ (OSHA-total); 5 mg/m³ (OSHA – resp); 5 mg/m³ (ACGIH, O. Reg. 833)
Silica (crystalline quartz): 10 mg/m³ (total dust); 3.3 mg/m³ (respirable) (OSHA); 0.05 mg/m³ (respirable - ACGIH); 0.1 mg/m³ (O. Reg. 845)

Use ventilation and dust collection to control exposure to below applicable limits.

Respiratory Protection: Wear NIOSH N-95 Dust Mask.

Eye Protection: Eye protection (chemical goggles, safety glasses and/or face shield) should be worn where there is a risk of hydrated lime exposure. Contact lenses should not be worn when working with lime products

Hand Protection: Use clean dry gloves

Skin Protection: Cover body with suitable clothes.

Refer to Ontario Regulation 845: Designated Substance – Silica.

9. PHYSICAL AND CHEMICAL PROPERTIES

<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Physical State</td>
<td>Solid</td>
</tr>
<tr>
<td>Odor &amp; Appearance</td>
<td>Odorless, white powder</td>
</tr>
<tr>
<td>pH</td>
<td>12.4 in saturated water solution at 25°C</td>
</tr>
<tr>
<td>Melting point</td>
<td>580°C</td>
</tr>
<tr>
<td>Boiling point</td>
<td>2850°C</td>
</tr>
<tr>
<td>Vapor pressure</td>
<td>Non volatile</td>
</tr>
<tr>
<td>Vapor density</td>
<td>Non volatile</td>
</tr>
<tr>
<td>Specific Gravity</td>
<td>2.24</td>
</tr>
<tr>
<td>Solubility</td>
<td>Slightly soluble in water: 0.2% @ 0°C</td>
</tr>
<tr>
<td></td>
<td>Soluble in acids, glycerin and sugar solutions</td>
</tr>
</tbody>
</table>

10. STABILITY AND REACTIVITY

<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Stability</td>
<td>Stable products, not very soluble.</td>
</tr>
<tr>
<td>Decomposition temperature</td>
<td>580°C, forms calcium oxide (CaO) and water</td>
</tr>
<tr>
<td>Reactivity</td>
<td>Reacts with acids to form calcium salts while generating heat.</td>
</tr>
<tr>
<td></td>
<td>Reacts with carbon dioxide in air to form calcium carbonate.</td>
</tr>
<tr>
<td>Conditions to avoid</td>
<td>Vicinity of incompatible materials</td>
</tr>
<tr>
<td>Incompatible materials</td>
<td>Acids; reactive fluoridated, brominated or phosphorous compounds; aluminum (may form hydrogen gas), reactive powdered metals; organic acid anhydrides; nitro-organic compounds; interhalogenated compounds</td>
</tr>
<tr>
<td>Hazardous decomposition products</td>
<td>Calcium oxide (CaO)</td>
</tr>
</tbody>
</table>
11. TOXICOLOGICAL INFORMATION

Toxicity: LD₅₀ oral (rat) for calcium hydroxide is 7340 mg/kg. This product is not listed by MSHA, OSHA, or IARC as a carcinogen, but this product may contain crystalline silica, which has been classified by IARC as (Group I) carcinogenic to humans when inhaled in the form of quartz or cristobalite. No reported Carcinogenicity, Reproductive Effects, Teratogenicity or Mutagenicity.

Exposure Limits: Refer to section 8.

Irritancy: Can cause severe irritation of eyes, skin, respiratory tract and gastrointestinal tract.

Chronic Exposure: Inhalation of silica can cause a chronic lung disorder, silicosis.

12. ECOLOGICAL INFORMATION

Alkaline substance that increases pH to a maximum of 12.4 in a saturated water solution at 25°C
Calcium hydroxide gradually reacts with CO₂ in air to form calcium carbonate (CaCO₃)
Calcium carbonate is ecologically neutral
Uncontrolled spillage in surface waters should be avoided since the increase pH could be detrimental to fish
Harmful to aquatic life in high concentration

13. DISPOSAL CONSIDERATIONS

Dispose according to federal, provincial/state and local environmental regulations.

14. TRANSPORTATION INFORMATION

Classification: TDG Not listed for ground transportation
               HMR Not listed for ground transportation

TDG: Transportation of Dangerous Goods Regulation (CAN)
HMR: Hazardous Materials Regulation (USA)
Product Name: HYDRATED LIME (continued)

15. REGULATORY INFORMATION

Symbol:

WHMIS RATING
D:A:E

NFPA RATING
HEALTH – 3 SPECIFIC HAZARD - ALK FLASH POINTS – 0 REACTIVITY – 1

HMIS RATING
HEALTH –3 SPECIFIC HAZARD - ALK FLASH POINTS – 0 REACTIVITY – 1

Risk Phrases:
Risk of serious damage to the eyes
Keep out of reach of children

Safety Phrases:
Keep storage container away from humidity
Avoid contact with skin and eyes. In case of contact with eyes, rinse immediately with water for at least 15 minutes

CPR (Canada):
This product has been classified in accordance with the hazard criteria of the Controlled Products Regulation (CPR) of Canada and this MSDS contains all information required by the CPR.

16. MISCELLANEOUS OTHER INFORMATION

Hydrated lime can be removed from objects (such as vehicles) using rags dampened with dilute vinegar. After applying dilute vinegar, vehicles (especially chrome surfaces) must be washed with water.

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